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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.043 wR factor = 0.147 Data-to-parameter ratio = 16.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 4-Methyl-N-(3-nitrophenyl)benzenesulfonamide

In the molecule of the title compound, m-(O<sub>2</sub>N)C<sub>6</sub>H<sub>4</sub>NHSO<sub>2</sub>-C<sub>6</sub>H<sub>4</sub>-p-Me or C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O<sub>4</sub>S, the planes of two benzene rings are orthogonal to each other [dihedral angle 92.4 (1)°].

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### Comment

The title molecule represents an important synthon for the preparation of a variety of sulfonamide drugs (Thakur *et al.*, 2005). The planes of the two benzene rings are orthogonal to each other [dihedral angle 92.4 (1)°; the C–N–S–C torsion angle in the central part of the molecule is -67.82 (17)°]. An N–H···O bond links the molecules into infinite chains running along the diagonal of the *ac* plane of the crystal structure.



### **Experimental**

The title compound was prepared according to the method of Sprague & Miller (1952). To a solution of 3-nitroaniline (27.6 g) in pyridine (100 ml) was added 4-tosyl chloride (41.8 g); the mixture was heated for 0.5 h at 373 K, then cooled to room temperature and poured into 500 ml of dilute HCl, yielding 4-methyl-*N*-(3-nitrophenyl)benzenesulfonamide in quantitative yield. X-ray quality crystals of the title compound (m.p. 405–407 K) were obtained by slow evaporation of an ethanol solution.

Crystal data	
$C_{13}H_{12}N_2O_4S$	$D_x = 1.482 \text{ Mg m}^{-3}$
$M_r = 292.31$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 10604
a = 12.766 (3) Å	reflections
b = 7.7327 (15) Å	$\theta = 3.0-27.5^{\circ}$
c = 13.550 (3)  Å	$\mu = 0.26 \text{ mm}^{-1}$
$\beta = 101.66 \ (3)^{\circ}$	T = 293 (2) K
V = 1310.0 (5) Å <sup>3</sup>	Block, light yellow
Z = 4	$0.29$ $\times$ 0.15 $\times$ 0.07 mm
Data collection	
Rigaku R-AXIS RAPID	2997 independent reflections
diffractometer	2519 reflections with $I > 2\sigma(I)$
Oscillation scans	$R_{\rm int} = 0.029$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(ABSCOR; Higashi, 1995)	$h = -16 \rightarrow 16$
$T_{\rm min} = 0.928, \ T_{\rm max} = 0.981$	$k = -10 \rightarrow 9$
12545 measured reflections	$l = -17 \rightarrow 16$

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#### Figure 1

Molecular structure of the title compound. Displacement ellipsoids are drawn at the 30% probability level; H atoms are shown as small spheres of arbitrary radii.



#### Figure 2

The packing of the title compound, projected on to the *ac* plane; the N– $H \cdots O$  bonds are shown as dashed lines. H atoms attached to C atoms have been omitted.

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.1119P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.043$	+ 0.1885P]
$wR(F^2) = 0.147$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} = 0.005$
2997 reflections	$\Delta \rho_{\rm max} = 0.39 \ {\rm e} \ {\rm \AA}^{-3}$
181 parameters	$\Delta \rho_{\rm min} = -0.44 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

Tab	le	1			
<b>a</b> 1					

Selected geometric parameters (Å,  $^\circ).$ 

S1-O4	1.4277 (14)	N1-O1	1.213 (2)
S1-O3	1.4341 (14)	N1-O2	1.227 (2)
S1-N2	1.6316 (16)	N1-C4	1.468 (2)
S1-C7	1.7628 (17)	N2-C6	1.416 (2)
O4-S1-O3	119.37 (9)	O3-S1-C7	109.24 (8)
O4-S1-N2	108.83 (8)	N2-S1-C7	106.61 (8)
O3-S1-N2	104.23 (9)	O1-N1-O2	122.59 (16)
O4-S1-C7	107.87 (9)		. ,
C7-S1-N2-C6	-67.82 (17)	O3-S1-C7-C12	67.70 (17)
S1-N2-C6-C1	161.38 (14)		

 Table 2

 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N2-H2A\cdots O2^{i}$	0.86	2.23	3.058 (2)	162
Symmetry code: (i) x	$-\frac{1}{2}, -y + \frac{1}{2}, z -$	- 1/2.		

All H atoms were positioned geometrically and refined using a riding model, with C–H = 0.93 and 0.96 Å, N–H = 0.86 Å and  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C}_{\rm aromatic},{\rm N})$  and  $1.5U_{\rm eq}({\rm C}_{\rm Me})$ .

Data collection: *RAPID-AUTO* (Rigaku, 1997); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXL97*.

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#### References

Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.

Rigaku (1997). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.

Sheldrick, G. M. (1997*a*). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

Sprague, J. M. & Miller, C. S. (1952). US Patent No. 2 608 506.

Thakur, M., Thakur, A., Khadikar, P. V. & Supuran, C. T. (2005). *Bioorg. Med. Chem. Lett.* **15**, 203–209.